

The only difference was the use of Et<sub>2</sub>O rather than petrol for oil separation from the aq. seed extract.

*Kinetic experiments* were carried out at 30° and pH 4.20 as follows: 2.0 ml of the acid lipase soln (1.0 mg/ml, 10 mM Tris buffer, pH 8.1) was mixed with 2.0 ml of the substrate emulsion (0.50 mmol/ml in 5% aq. gum arabic). Enough 0.1 M HOAc was then added to adjust the pH to 4.20. The amount of free acid generated was followed by potentiometric titration with 50 mM NaOH as described in ref. [8]. Initial reaction rates were determined according to Boeker [9] as:

$$v_i = \lim_{P \rightarrow 0} \Delta P / \Delta t.$$

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## FURTHER SILPHINENE DERIVATIVES FROM *CINERARIA GEIFOLIA* VAR. *GLABRA*

J. JAKUPOVIC, W.-R. ABRAHAM and F. BOHLMANN

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany

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**Key Word Index**—*Cineraria geifolia* var. *glabra*; Compositae; sesquiterpenes; silphinene derivatives.

**Abstract**—A reinvestigation of the sesquiterpenes from the aerial parts of *Cineraria geifolia* var. *glabra* afforded eight new silphinene derivatives. One of these compounds had been isolated previously but its structure had not been elucidated with certainty. The stereochemistry of a corresponding isovalerate from *Callilepis salicifolia* has to be revised.

### INTRODUCTION

A keto diester had been isolated from *Cineraria geifolia* var. *glabra* (tribe Senecioneae) [1]. However, the decision between two possible structures was difficult and therefore only a preliminary structure was reported [1]. A reinvestigation of this compound and some minor related sesquiterpenes allowed a clear assignment of the structures. In addition to the acetoxy angelate **1**, which was isolated previously, three further diesters (**2–4**), the hydroxy ester **5** and the monoesters **6–8** were obtained.

### RESULTS AND DISCUSSION

The <sup>1</sup>H NMR spectral data of **1** (Table 1) were close to those of a ketone obtained by oxidation of a hydroxy isovalerate from *Callilepis salicifolia* [2]. Careful spin decoupling allowed the assignment of all signals. The resulting sequences of the carbons C-1–C-3 and C-7, C-11, C-10–C-9 (C-15) showed that we were dealing with a

derivative of silphinene [3]. The stereochemistry, however, could only be established by NOE difference spectroscopy (Table 2). Also the relative position of the ester group could be assigned from the NOEs. Thus irradiation of H-12 caused a clear effect with the methyl signals of the angelate residue, while H-13 and H-14 showed NOEs with the acetoxy methyl. Furthermore, inspection of a model indicated that due to the quasi-axial orientation of the 11β-acetoxy group the H-5 signal was shifted downfield. As H-11 showed a clear NOE with H-7 and H-10α the β-orientation of the acetoxy group was established. The assignment of H-10α followed from the observed small coupling with H-9. As H-15 showed an NOE with H-1 the α-orientation of H-9 was settled. Similarly all other orientations followed from the observed NOEs. The <sup>13</sup>C NMR signals (see Experimental) also agreed nicely with the proposed structure.

The <sup>1</sup>H NMR spectra of **2–4** (Table 1) were nearly identical with that of **1**, only the typical signal of the

Table 1.  $^1\text{H}$  NMR spectral data of 1–9 (400 MHz,  $\text{CDCl}_3$ , TMS as int. standard)

H	1	2	3	4	5	6	7	8	9
1	7.57 <i>d</i>		7.57 <i>d</i>		7.56 <i>d</i>	7.58 <i>d</i>	7.57 <i>d</i>		7.56 <i>d</i>
2	6.12 <i>d</i>		6.12 <i>d</i>		6.09 <i>d</i>	6.01 <i>d</i>	6.01 <i>d</i>		6.00 <i>d</i>
5	5.43 <i>s</i>		5.33 <i>s</i>		5.45 <i>s</i>	5.15 <i>s</i>	5.03 <i>s</i>		5.04 <i>s</i>
7	2.22 <i>d</i>	2.23 <i>d</i>	2.21 <i>d</i>		2.04 <i>d</i>	2.14 <i>dd</i>	2.10 <i>dd</i>		2.11 <i>dd</i>
9	2.62 <i>ddq</i>		2.62 <i>ddq</i>		2.82 <i>ddq</i>	2.19 <i>ddq</i>	2.18 <i>ddq</i>	2.19 <i>ddq</i>	2.19 <i>ddq</i>
10 $\alpha$	1.57 <i>ddd</i>		1.56 <i>ddd</i>		1.56 <i>ddd</i>	1.34 <i>dddd</i>	1.35 <i>m</i>		1.34 <i>dddd</i>
10 $\beta$	2.10 <i>dd(br)</i>		2.09 <i>dd(br)</i>		1.92 <i>dd(br)</i>	1.95 <i>m</i>	1.95 <i>m</i>		1.94 <i>dd(br)</i>
11 $\alpha$	5.25 <i>dd(br)</i>		5.23 <i>dd(br)</i>		4.36 <i>dd(br)</i>	1.80 <i>m</i>	1.79 <i>m</i>		1.78 <i>dd(br)</i>
11 $\beta$	—		—		—	1.43 <i>dddd</i>	1.40 <i>m</i>		1.42 <i>dddd</i>
12	0.90 <i>s</i>	0.89 <i>s</i>	0.87 <i>s</i>		0.87 <i>s</i>	0.84 <i>s</i>	0.81 <i>s</i>		0.82 <i>s</i>
13	0.94 <i>s</i>	0.93 <i>s</i>	0.92 <i>s</i>		1.16 <i>s</i>	0.94 <i>s</i>	0.91 <i>s</i>	0.90 <i>s</i>	0.91 <i>s</i>
14	1.25 <i>s</i>	1.22 <i>s</i>	1.21 <i>s</i>	1.22 <i>s</i>	1.24 <i>s</i>	1.20 <i>s</i>	1.17 <i>s</i>	1.16 <i>s</i>	1.16 <i>s</i>
15	0.92 <i>d</i>		0.92 <i>d</i>		0.92 <i>d</i>	0.92 <i>d</i>	0.91 <i>d</i>		0.92 <i>d</i>
OAc	2.17 <i>s</i>	2.18 <i>s</i>	2.15 <i>s</i>		—	—	—	2.13 <i>s</i>	—
OR	6.11 <i>qq</i>	6.99 <i>dq</i>	2.67 <i>qq</i>	2.49 <i>dq</i>	6.09 <i>qq</i>	6.10 <i>qq</i>	2.48 <i>dq</i>		2.32 <i>dd</i>
	2.04 <i>dq</i>	1.83 <i>dq</i>	1.26 <i>d</i>	2.40 <i>dq</i>	2.03 <i>dq</i>	2.03 <i>dq</i>	2.40 <i>dq</i>		2.27 <i>dd</i>
	1.94 <i>dq</i>	1.88 <i>dq</i>	1.23 <i>d</i>	1.19 <i>t</i>	1.94 <i>dq</i>	1.94 <i>dq</i>	1.19 <i>t</i>		2.17 <i>ddqq</i>
									1.00 <i>d</i>
									0.99 <i>d</i>

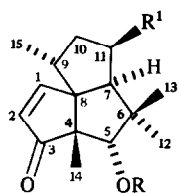
*J* (Hz): Compounds 1–5: 1, 2 = 5.5; 7, 11 = 4; 9, 10 $\alpha$  = 10 $\alpha$ , 10 $\beta$  = 12.5; 9, 10 $\beta$  = 6; 9, 15 = 7; 10 $\alpha$ , 11 = 3; compounds 6–9: 1, 2 = 5.5; 7, 11 $\alpha$  = 8; 7, 11 $\beta$  = 9, 10 $\alpha$  = 10 $\alpha$ , 10 $\beta$  = 12; 9, 10 $\beta$  = 10 $\beta$ , 11 $\beta$  = 6; 9, 15 = 10 $\alpha$ , 11 $\alpha$  = 7; 10 $\alpha$ , 11 $\beta$  = 5; 11 $\alpha$ , 11 $\beta$  = 12.5; OAng, OTigl: 3', 4' = 7; 3', 5' = 4', 5' = 1.5; OiBu: 2', 3' = 2', 4' = 7; OProp: 2', 2' = 16.5; 2, 3 = 7.

changed ester residue being different. A small upfield shift of the H-5 signal in the isobutyrate **3** and the propionate **4** is characteristic for saturated esters if compared with the shift of unsaturated ones. This also allowed the assignment of the relative position of the ester group in **2**, while biogenetic considerations led to the assumption that in **3** and **4** the *O*-acetate group also was at C-11. This was supported by the fact that in all cases (also with **5–9**, see below) a strong MS fragment was observed formed by loss

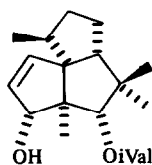
of RCO, which most likely was due to a neighbouring group effect of the cyclopentenone system.

The  $^1\text{H}$  NMR spectrum of **5** (Table 1) only differed in the chemical shift of H-11 from that of **1**. Accordingly, the corresponding desacetyl derivative was present.

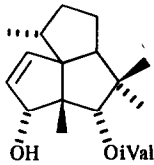
The  $^1\text{H}$  NMR spectra of **6–8** (Table 1) were nearly identical with that of the already mentioned isovalerate (**9**) obtained by oxidation of the *Callilepis* carbinol [2]. Again all signals could be assigned by spin decoupling and the



	1	2	3	4	5	6	7	8	9
R	Ang	Tigl	iBu	Prop	Ang	Ang	Prop	Ac	iVal
R <sup>1</sup>	OAc	OAc	OAc	OAc	OH	H	H	H	H



10



11

Table 2. NOE effects with **1** and **9**

Irradiation of	NOE observed (effects in % in parentheses)
<b>1</b>	
H-12	H-1 (2); H-2 (2); H-7 (10); OAng: H-4' (1); H-5' (1)
H-15	H-1 (6); H-10 $\beta$ (2)
H-13	H-5 (8); H-11 (6); OAc (2)
H-14	H-5 (12); H-9 (14); OAc (2)
H-10 $\alpha$	H-11 (5); H-7 (3)
H-10 $\beta$	H-9 (6)
H-7	H-1 (5); H-11 (10); H-10 $\beta$ (3); H-12 (5)
H-9	H-10 $\beta$ (6); H-14 (5)
H-11	H-7 (10); H-10 $\alpha$ (4); H-13 (4)
H-5	H-13 (4); H-14 (5)
<b>9</b>	
H-12	H-1 (2); H-2 (2); H-7 (10); H-13 (10)
H-13	H-5 (6); H-11 (4); H-12 (6); H-14 (4)
H-14	H-5 (7); H-9 (10); H-13 (4)
H-15	H-1 (6)

stereochemistry was established by NOE difference spectroscopy. In the case of **9**, especially clear NOEs between H-14 and H-5 and between H-13 and H-5 showed that the configuration at C-5 was the same as in **1-5**. The same effects were obtained with the esters **6-8**. Thus the stereochemistry of all the monoesters was settled. This required a revision of the configuration of the silphenene derivative from *Callilepis salicifolia* (**10**) to **11**, the epimer at C-3 and C-5. Accordingly, reduction of **9** gave the natural compound. As followed from inspection of a model, the hydride attack is probably directed by the configuration of the ester group at C-5. From the observed Cotton-effect of **9** the opposite absolute configuration was deduced. Biogenetic considerations, however, showed that probably all these triquinanes are derived from (–)-caryophyllene [**3**] indicating that the presented absolute configuration is more likely. This is supported by the fact that (–)-caryophyllene occurs together with all the known sesquiterpenes of this type [**3**]. In one case this could be established by synthesis of (–)-silphiperfol-6-ene starting with (R)-(+)-pulegone [**4**].

#### EXPERIMENTAL

The extract (Et<sub>2</sub>O–petrol, 1:2) of the air dried aerial parts (10 g, collected on Table Mountain in September 1977, voucher 77/337) was separated in the usual fashion [**5**]. TLC (silica gel PF 254, Et<sub>2</sub>O–petrol–CHCl<sub>3</sub>, 1:7:2, four developments) gave 2 mg **6** (*R<sub>f</sub>* 0.73, values always for four developments), 3 mg **7** (*R<sub>f</sub>* 0.64), 3 mg **8** (*R<sub>f</sub>* 0.60), 15 mg **1** (*R<sub>f</sub>* 0.38), 2 mg **2** (*R<sub>f</sub>* 0.34), 2 mg **3** (*R<sub>f</sub>* 0.32), 5 mg **4** (*R<sub>f</sub>* 0.27) and 2 mg **5** (*R<sub>f</sub>* 0.20). Though all compounds were homogeneous by TLC in different solvent mixtures and by 400 MHz <sup>1</sup>H NMR spectroscopy they could not be induced to crystallize and they were obtained as colourless oils.

**11β-Acetoxy-5α-angeloyloxy-silphenen-3-one (1).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1745 (OAc), 1715 (C=CCO<sub>2</sub>R, C=CC=O); MS *m/z* (rel. int.): 374.209 [M]<sup>+</sup> (5) (calc. for C<sub>22</sub>H<sub>30</sub>O<sub>5</sub>: 374.209), 291 [M–COR]<sup>+</sup> (32), 274 [M–RCO<sub>2</sub>H]<sup>+</sup> (12), 232 [274–ketene]<sup>+</sup> (5), 214 [274–HOAc]<sup>+</sup> (20), 83 [C<sub>4</sub>H<sub>7</sub>CO]<sup>+</sup> (100), 55 [83–CO]<sup>+</sup> (62), 43 [MeCO]<sup>+</sup> (42); <sup>13</sup>C NMR (CDCl<sub>3</sub>, C-1–C-11): 167.6 d, 138.3 d, 211.0 s, 64.1 s, 86.4 d, 42.6 s, 61.9 d, 57.2 s, 35.2 d, 41.9 t, 76.2 d; OAc: 168.0 s, 128.2 s, 131.9 d; OAc: 170.0 s (methyl signals: 26.6 q, 23.8 q, 21.7 q, 20.6 q, 19.9 q, 15.9 q, 15.8 q);

$$[\alpha]_{25}^{25} = \frac{589}{-69} \frac{578}{-72} \frac{546}{-85} \frac{436 \text{ nm}}{-184} (\text{CHCl}_3; c = 0.7).$$

**11β-Acetoxy-5α-tigloyloxy-silphenen-3-one (2).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1745 (OAc), 1715 (C=CCO<sub>2</sub>R, C=CC=O); MS *m/z* (rel. int.): 374.209 [M]<sup>+</sup> (3) (calc. for C<sub>22</sub>H<sub>30</sub>O<sub>5</sub>: 374.209), 291 [M–COR]<sup>+</sup> (28), 275 [M–OCOR]<sup>+</sup> (10), 215 [275–HOAc]<sup>+</sup> (22), 83 [C<sub>4</sub>H<sub>7</sub>CO]<sup>+</sup> (100).

**11β-Acetoxy-5α-isobutyryloxy-silphenen-3-one (3).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1740 (OAc, CO<sub>2</sub>R), 1715 (C=CC=O); MS *m/z* (rel. int.): 362.209 [M]<sup>+</sup> (1) (calc. for C<sub>21</sub>H<sub>30</sub>O<sub>5</sub>: 362.209), 291 [M–COR]<sup>+</sup> (30), 275 [M–OCOR]<sup>+</sup> (9), 215 [275–HOAc]<sup>+</sup> (21), 71 [C<sub>3</sub>H<sub>7</sub>CO]<sup>+</sup> (30), 43 [71–CO]<sup>+</sup> (100).

**11β-Acetoxy-5α-propionyloxy-silphenen-3-one (4).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1740 (OAc, CO<sub>2</sub>R), 1715 (C=CC=O); MS *m/z* (rel. int.): 348.194 [M]<sup>+</sup> (10) (calc. for C<sub>20</sub>H<sub>28</sub>O<sub>5</sub>: 348.194), 292 [M–O=C=CHMe]<sup>+</sup> (42), 291 [M–COEt]<sup>+</sup> (85), 250 [292–ketene]<sup>+</sup> (27), 231 [291–HOAc]<sup>+</sup> (37), 57 [C<sub>2</sub>H<sub>5</sub>CO]<sup>+</sup> (100).

**11β-Hydroxy-5α-angeloyloxy-silphenen-3-one (5).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 3600 (OH), 1715 (C=CCO<sub>2</sub>R, C=CC=O); MS *m/z* (rel. int.): 332.199 [M]<sup>+</sup> (1) (calc. for C<sub>20</sub>H<sub>28</sub>O<sub>4</sub>: 332.199), 249 [M–COR]<sup>+</sup> (21), 233 [M–OCOR]<sup>+</sup> (11), 215 [233–H<sub>2</sub>O]<sup>+</sup> (37), 83 [C<sub>4</sub>H<sub>7</sub>CO]<sup>+</sup> (100), 55 [83–CO]<sup>+</sup> (67).

**5α-Angeloyloxy-silphenen-3-one (6).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1715 (C=CCO<sub>2</sub>R, C=CC=O); MS *m/z* (rel. int.): 316.204 [M]<sup>+</sup> (10) (calc. for C<sub>20</sub>H<sub>28</sub>O<sub>3</sub>: 316.204), 233 [M–COR]<sup>+</sup> (90), 217 [M–OCOR]<sup>+</sup> (55), 83 [C<sub>4</sub>H<sub>7</sub>CO]<sup>+</sup> (100).

**5α-Propionyloxy-silphenen-3-one (7).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1740 (CO<sub>2</sub>R), 1715 (C=CC=O); MS *m/z* (rel. int.): 290.188 [M]<sup>+</sup> (20) (calc. for C<sub>18</sub>H<sub>26</sub>O<sub>3</sub>: 290.188), 233 [M–COR]<sup>+</sup> (100), 217 [M–OCOEt]<sup>+</sup> (60), 57 [EtCO]<sup>+</sup> (70).

**5α-Acetoxy-silphenen-3-one (8).** IR  $\nu_{\text{max}}^{\text{CCl}_4}$  cm<sup>–1</sup>: 1745 (OAc), 1715 (C=CC=O); MS *m/z* (rel. int.): 276.172 [M]<sup>+</sup> (17) (calc. for C<sub>17</sub>H<sub>24</sub>O<sub>3</sub>: 276.172), 233 [M–COMe]<sup>+</sup> (100), 217 [M–OAc]<sup>+</sup> (65).

**Borane reduction of 5α-isovaleryloxy-silphenen-3-one.** To 5 mg **9** in 2 ml MeOH, 10 mg NaBH<sub>4</sub> was added. After 15 min. dil H<sub>2</sub>SO<sub>4</sub> was added. TLC (Et<sub>2</sub>O–petrol, 1:1) of the reaction product gave 3 mg 3α-hydroxy-5α-isovaleryloxy-silphenene, identical with the natural compound by co-TLC and <sup>1</sup>H NMR.

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